Ab initio wavefunction based methods for excited states in solids: correlation corrections to the band structure of ionic oxides

L. Hozoi, U. Birkenheuer, and P. Fulde Max-Planck-Institut für Physik komplexer Systeme, Nöthnitzer Str. 38, 01187 Dresden, Germany

> A. Mitrushchenkov and H. Stoll Universität Stuttgart, Pfaffenwaldring 57, 70550 Stuttgart, Germany (Dated: February 5, 2008)

Ab initio wavefunction based methods are applied to the study of electron correlation effects on the band structure of oxide systems. We choose MgO as a prototype closed-shell ionic oxide. Our analysis is based on a local Hamiltonian approach and performed on finite fragments cut from the infinite solid. Localized Wannier functions and embedding potentials are obtained from prior periodic Hartree-Fock (HF) calculations. We investigate the role of various electron correlation effects in reducing the HF band gap and modifying the band widths. On-site and nearest-neighbor charge relaxation as well as long-range polarization effects are calculated. Whereas correlation effects are essential for computing accurate band gaps, we found that they produce smaller changes on the HF band widths, at least for this material. Surprisingly, a broadening effect is obtained for the O 2p valence bands. The ab initio data are in good agreement with the energy gap and band width derived from thermoreflectance and x-ray photoemission experiments. The results show that the wavefunction based approach applied here allows for well controlled approximations and a transparent identification of the microscopic processes which determine the electronic band structure.

I. INTRODUCTION

The proper treatment of electron correlation effects in molecules and solids stands at the heart of modern electronic structure theory^{1,2,3,4,5}. In the study of molecular systems at least, wavefunction based quantum chemistry is known to provide a rigorous theoretical framework for addressing the electron correlation problem. The standard quantum chemical methods^{4,5} make possible the construction of approximate wavefunctions at levels of increasing sophistication and accuracy and offer thus a systematic route to converged results. Advanced wavefunction based calculations can be routinely performed nowadays for small and medium size molecules. However, algorithms able to treat electron correlation effects in periodic systems are still at their infancy. The simplest correlation method is based on the Møller-Plesset equations and second-order Møller-Plesset (MP2) perturbational schemes for solids were actually implemented by several groups 6,7,8,9,10,11,12,13,14 . As an extension of MP2, ab initio many-body Green's function techniques were developed too 15,16,17,18,19. In addition, investigations based on coupled-cluster (CC) theory were initiated 13,20,21,22,23 .

The main point when applying quantum chemical methods to correlation calculations is to make use of the local character of the correlation hole which is surrounding an electron. The latter optimizes the Coulomb repulsion between electrons and its accurate description is the essence of the correlation treatment. Starting point is a Hartree-Fock (HF) calculation on top of which the correlation calculations are implemented. For insulators and semiconductors, the development of efficient computational tools is facilitated by the use of optimally localized Wannier functions. Several orbital localization procedures were proposed in the context of periodic HF cal-

culations. They normally rest on the *a posteriori* transformation of the crystal Bloch orbitals, see for example Refs. 6,14,20,21,22,24,25. Alternative approaches were elaborated by Shukla *et al.*^{26,27,28,29} and in the Toulouse group^{11,30}, where the self-consistent-field (SCF) equations are solved directly in the Wannier representation.

Local correlation methods such as the incremental scheme of $Stoll^{31,32}$ and the local Hamiltonian formalism of Horsch $et~al.^{33}$ and Gräfenstein $et~al.^{34,35,36}$ were applied before to the rigorous determination of both ground-state^{13,26,27,28} and excited-state^{18,37,38,39,40} properties of infinite systems. Materials under investigation were crystalline LiH, LiF, and LiCl^{18,26,27,28}. diamond^{38,40}, silicon³⁸, beryllium³⁹, carbon, and boronnitrogen chains¹³, and *trans*-polyacetylene³⁷. In the present paper we extend these studies to the case of ionic oxide compounds and choose MgO as a prototype and, at the same time, relatively simple insulating oxide. We compute correlation induced corrections to the valence and conduction HF energy bands and find that our final estimate for the fundamental gap is in rather good agreement with the experimental data. Our analysis provides also a clear picture of the major correlation effects that are responsible for the reduction of the HF band gap. The present study on magnesium oxide should open the way to similar calculations for more complex materials such as the 3d transition-metal oxides.

II. COMPUTATIONAL APPROACH

Periodic HF calculations were carried out for crystalline MgO with the CRYSTAL package⁴¹. Wannier functions associated with the HF valence and low-lying conduction bands were determined with the Wannier-

Boys localization module²⁴ of the same program. For conduction-band states, this may be a somewhat tedious task. However, localized Wannier orbitals can also be derived in more difficult cases with band disentanglement techniques²⁵.

For computing the correlation induced corrections to the valence and conduction bands we adopt the same quasiparticle picture and local Hamiltonian formalism as employed in Refs. 33.34.35.36.37.38.39.40. The physical reasoning is as follows. When an electron (or hole) is added to the N-particle system, its surroundings relax and polarize due to the additional charge. This response of the system lowers the energy it takes to add the extra electron (or hole). The particle plus the modified surroundings move together through the system in the form of a Bloch wave and define a quasiparticle. At the same time, some of the correlation contributions which are present in the ground-state of the N-particle system are no longer operative in the (N+1) [or (N-1)] -particle system, because some excitations are now blocked. This effect is called loss of ground-state correlation and must be accounted for too^{33,38}. A basic feature of our approach is that the correlation treatment is performed on a finite cluster \mathcal{C} cut from the extended periodic system. With the cluster-in-solid embedding technique elaborated by Birkenheuer et al. 38,42,43 , the cluster \mathcal{C} is divided into an active region C_A that supports the local occupied and virtual orbitals entering the post-HF calculations and a spatial "buffer" domain C_B including a number of atomic sites (and basis functions) whose role is to provide a good representation for the tails of the Wannier-like orbitals of the active region \mathcal{C}_A . Thus $\mathcal{C} = \mathcal{C}_A + \mathcal{C}_B$. All Wannier orbitals centered in C_B (and also the environment) will be held frozen in the correlation calculations.

The orbital set associated with the finite-size cluster is obtained from the data supplied by the Wannier-Boys localization module of CRYSTAL²⁴. The interface program⁴⁴ written for this purpose yields in addition an embedding potential corresponding to the frozen environment, i.e., the surrounding HF electron sea. The detailed procedure for constructing the embedded cluster and the associated orbital set is described elsewhere^{42,43}. Here we merely give an outline of this. The core, valence, and low-lying conduction-band cluster orbitals are generated by projecting the crystal Wannier functions onto the set of atomic basis functions attached to the cluster region \mathcal{C} ,

$$|w'_{n}(\mathbf{R})\rangle = \sum_{\beta,\alpha\in\mathcal{C}} |\beta\rangle S_{\beta\alpha}^{-1} \langle \alpha|w_{n}(\mathbf{R})\rangle,$$
 (1)

where α and β are Gaussian basis functions centered within the region \mathcal{C} , $S_{\beta\alpha}^{-1}$ represents the inverse overlap matrix for the basis set attached to \mathcal{C} , $|w_n(\mathbf{R})\rangle$ is a Wannier orbital with index n and centered in the unit cell corresponding to the lattice vector \mathbf{R} , and $|w_n'(\mathbf{R})\rangle$ is its projected counterpart expressed exclusively in terms of basis functions centered in \mathcal{C} . The $|w_n'(\mathbf{R})\rangle$ functions are neither normalized nor are they orthogonal to

each other because of the projection procedure mentioned above. Therefore they are group-wise orthonormalized in the following order: active core, active occupied, active low-lying conduction-band orbitals, buffer core, buffer occupied, buffer low-lying conduction-band orbitals. This way the contamination of the most important types of orbitals is minimized. Group-wise orthonormalization actually means to use Schmidt orthogonalization for the inter-group orthogonalization and Löwdin orthonormalization inside the groups. This set of orthonormal orbitals will be denoted by $|\tilde{w}_n'(\mathbf{R})\rangle$. For the construction of the variational space to be used in the subsequent correlation calculations we follow the prescription suggested by Pulay and Saebø⁴⁵. So-called virtual projected atomic orbitals (PAO's) are generated from the Gaussian basis functions associated with the active region C_A by projecting out the occupied and the low-lying conductionband orbitals $|\tilde{w}_n'(\mathbf{R})\rangle$ via a Schmidt orthogonalization scheme. Thereafter, the PAO's are Löwdin orthonormalized among themselves to facilitate their subsequent use in the correlation calculations.

By making the buffer region C_B sufficiently large the original Wannier orbitals of \mathcal{C}_A are well represented. For example, for the Wannier orbitals corresponding to the lowest four conduction bands of MgO and centered at a Mg site, the six ligands surrounding the given Mg ion should be included in C_A because large contributions arise not only from the Mg 3s and 3p basis functions but also from the nearest-neighbor (NN) oxygen 2s,3s and 2p,3pcomponents. Farther neighbors need not be included in the central region \mathcal{C}_A but may be put into the buffer zone C_B because the weight of the longer-range tails is small. Whereas a high-quality description can be easily achieved for the Wannier orbitals centered in the active region, the representation of the Wannier functions in the buffer zone is less accurate. The impact of this deficiency on the correlation calculations, however, is completely compensated by an appropriate choice of the embedding potential. Actually, the Gaussian orbital representation $V_{\alpha\beta}^{\text{emb}}$ of the embedding potential is constructed by

$$V_{\alpha\beta}^{\text{emb}} = F_{\alpha\beta}^{\text{crys}} - F[P^{\mathcal{C}}]_{\alpha\beta}, \qquad \alpha, \beta \in \mathcal{C}$$
 (2)

where F^{crys} is the self-consistent Fock operator from the periodic HF calculation and $F[P^{\mathcal{C}}]$ is the Fock operator associated with the density operator

$$P^{\mathcal{C}} = 2\sum_{\nu}^{\text{occ}} |\tilde{w}_{\nu}'\rangle\langle\tilde{w}_{\nu}'| \tag{3}$$

arising from all occupied orbitals $|\tilde{w}'_{\nu}\rangle$ which enter the subsequent correlation calculations explicitly. This way, the correlation calculations are effectively performed in an *infinite* frozen HF environment.

The data concerning the occupied and virtual orbitals of the cluster is transferred via the CRYSTAL-MOLPRO interface⁴⁴ to the quantum chemistry program MOLPRO⁴⁶. The same holds for the matrix representation $F_{\alpha\beta}^{\text{crys}}$ of the self-consitent Fock operator of the periodic

host system. The embedding potential itself is constructed according to Eqs. (2) and (3) using the MATROP module of the MOLPRO program package.

Local electron removal and electron addition oneparticle configurations can be defined in terms of the set of occupied and virtual orbitals localized within the spatial domain C_A :

$$|\Phi_{\mathbf{R}\rho\sigma}^{N-1}\rangle = c_{\mathbf{R}\rho\sigma}|\Phi\rangle$$
 and $|\Phi_{\mathbf{R}q\sigma}^{N+1}\rangle = c_{\mathbf{R}q\sigma}^{\dagger}|\Phi\rangle$, (4)

where $c_{\mathbf{R}p\sigma}$ and $c_{\mathbf{R}q\sigma}^{\dagger}$ are annihilation and creation operators for the valence and conduction band σ -spin orbitals $|\tilde{w}'_p(\mathbf{R})\rangle$ and $|\tilde{w}'_q(\mathbf{R})\rangle$, respectively, and $|\Phi\rangle$ is the single-determinant ground-state wavefunction of the N-electron system. For clusters which are large enough, the Hartree-Fock valence and conduction energy bands of the periodic crystal can be recovered by diagonalizing \mathbf{k} -dependent matrices of the following form:

$$H_{nn'}^{\rm HF}(\mathbf{k}) = \sum_{\mathbf{R}} e^{i\mathbf{k}\mathbf{R}} \langle \Phi_{\mathbf{0}n\sigma}^{N\mp1} | H - E_0^{\rm HF} | \Phi_{\mathbf{R}n'\sigma}^{N\mp1} \rangle. \quad (5)$$

Here $\mathbf{0}$ stands for the reference unit cell and E_0^{HF} denotes the ground-state HF energy of the neutral N-electron system. The diagonal terms $\mathbf{R} = \mathbf{0}$, n' = n in this expression are directly related to the on-site excitation energies within the Koopmans approximation^{4,5}, i.e., ionization potentials

$$IP_{pp}^{HF}(\mathbf{0}) = \langle \Phi_{\mathbf{0}p\sigma}^{N-1} | H | \Phi_{\mathbf{0}p\sigma}^{N-1} \rangle - E_0^{HF} = -\epsilon_{\mathbf{0}p}^{HF} > 0 \quad (6)$$

and electron affinities

$$\mathrm{EA}_{qq}^{\mathrm{HF}}(\mathbf{0}) = E_0^{\mathrm{HF}} - \langle \Phi_{\mathbf{0}q\sigma}^{N+1} | H | \Phi_{\mathbf{0}q\sigma}^{N+1} \rangle = -\epsilon_{\mathbf{0}q}^{\mathrm{HF}}, \quad (7)$$

with the latter being negative in the case of MgO. The off-diagonal terms are the hopping matrix elements in a tight-binding representation,

$$t_{nn'}^{\rm HF}({\bf R}) = \langle \Phi_{{\bf 0}n\sigma}^{N\mp 1} | H - E_0^{\rm HF} | \Phi_{{\bf R}n'\sigma}^{N\mp 1} \rangle .$$
 (8)

Starting from the Hartree-Fock energy bands we include next the effects of electron correlations. Within the quasiparticle approximation we may introduce for the (N+1)-particle system a wavefunction of the following form:

$$|\Psi_{\mathbf{R}q\sigma}^{N+1}\rangle = \exp(S) c_{\mathbf{R}q\sigma}^{\dagger} |\Psi\rangle,$$
 (9)

where $|\Psi\rangle$ is the ground-state of the N electron system. The operator S does not commute with $c_{\mathbf{R}q\sigma}^{\dagger}$ and describes the correlation hole. For a more detailed discussion we refer to Ref. 1. Alternatively, we may write $|\Psi_{\mathbf{R}q\sigma}^{N+1}\rangle = \Omega\,c_{\mathbf{R}q\sigma}^{\dagger}|\Phi\rangle$, where $|\Phi\rangle$ is the SCF ground-state and Ω acts like a wave- or Moeller operator³. Therefore, when correlations are taken into account, the analogue of (5) becomes

$$H_{nn'}(\mathbf{k}) = \sum_{\mathbf{R}} e^{i\mathbf{k}\mathbf{R}} \langle \Psi_{\mathbf{0}n\sigma}^{N\mp 1} | H - E_0 | \Psi_{\mathbf{R}n'\sigma}^{N\mp 1} \rangle, \qquad (10)$$

where E_0 is the energy of the N-particle correlated ground-state $|\Psi\rangle$ (see also Refs. 34,35,36,37,38). The correlation hole around an added electron (or hole) consists of a short-range and a long-range part. The short-range part originates from intra-atomic and short-range interatomic relaxation and polarization effects. We construct the short-range part of the correlation hole by separate orbital optimizations. Thereby the Wannier orbital to which the extra electron (hole) is attached is kept frozen and the changes within a finite region around it are determined by performing an additional SCF calculation^{38,39}. This is sufficient if electron correlations are weak or moderate. If they are strong we would have to take into account the fact that changes in the nearby surroundings are decreased when correlation effects among the electrons in that neighborhood are accounted for. The long-range part of the correlation hole consists of long range polarization of the environment. The effect of longrange polarization on the diagonal Hamiltonian matrix elements is estimated in this paper by applying the approximation of a dielectric continuum. With the known dielectric constant of MgO, $\epsilon_0 = 9.7$, we can directly determine the polarization energy of a charge $\pm e$ outside a sphere of radius R. In addition, differential correlation effects related to the existence of a different number of electrons in the system's ground-state and in the $(N\pm 1)$ excited states are investigated by subsequent configurationinteraction (CI) calculations.

III. CORRELATION CORRECTIONS TO THE BAND STRUCTURE OF MGO

Magnesium oxide is a prototype closed-shell ionic material that crystallizes under normal conditions in the rocksalt structure. It is extensively used in materials science as substrate for the epitaxial growth of films of other compounds. In recent years, it has attracted renewed in-

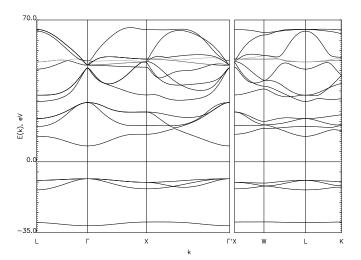


FIG. 1: Hartree-Fock band structure of bulk MgO. The core O 1s and Mg 1s - 2p bands are not shown in the figure.

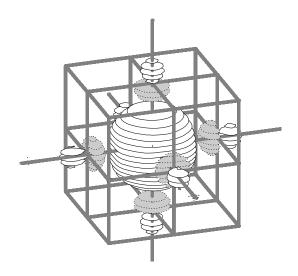


FIG. 2: Plot of the Mg 3s-like conduction-band Wannier orbital after projection onto a [Mg₁₉O₁₄] cluster. There is substantial weight at the NN oxygen sites.

terest because of its use as tunnel barrier in magnetic tunnel junctions 47 . A different area where MgO attracted attention is optoelectronics. It turns out that when alloyed with ZnO, depending on the precise chemical composition, the band gap of the system can be tuned along an interval ranging from 3.3 to 7.8 eV⁴⁸. The latter number represents the fundamental gap of MgO⁴⁹.

The valence bands of MgO have oxygen 2p character, whereas the lowest conduction bands are mainly related to the Mg 3s and 3p orbitals, with some admixture from the O 2s,3s and 2p,3p functions. The Hartree-Fock energy bands are shown in Fig.1. We employed for our calculations the lattice constant reported by Sasaki et al., $a=4.217\,\text{Å}^{50}$, and Gaussian-type basis sets from the standard CRYSTAL library. Basis sets of triple-zeta quality, Mg 8-511⁵¹, were used for the cations and triple-zeta basis sets suplemented with polarization functions, O 8-411*51, were applied for the more polarizable O ions. At the Hartree-Fock level and with this choice of the basis functions, the fundamental gap of the system is 16.20 eV, i.e., 8.4 eV larger than the experimental value. As illustrated in Fig.1, there is a separation of about 16 eV between the O 2s and O 2p bands. There is also a clear separation between the low-lying Mg 3s, 3p complex and the other conduction bands. Density-functional calculations within the local density approximation and using the same Gaussian basis sets as for the HF calculations give a gap of 5.0 eV between the valence and conduction bands, nearly 3 eV lower than observed in experiment.

Projected Wannier orbitals associated with the Mg 3s, 3p conduction-band complex are plotted in Fig.2 and Fig.3. We note that the Wannier-Boys localization module of the CRYSTAL program yields a set of Mg sp^3 hybrids for the lowest four conduction-band states. In order to arrive at a set of s and x, y, z-like functions, we applied to those projected hybrids a Pipek-Mezey localization

procedure 52 , as implemented in the MOLPRO package 46 . In this localization scheme, the number of atomic orbitals contributing to a given molecular-like composite is minimized. Whereas the Mg 3s and 3p Wannier functions have substantial weight at the NN ligand sites, the valence-band O 2p (and 2s) Wannier orbitals are much more compact and contributions from neighboring ions are not visible in plots like those shown in Figs. 2 and 3. In the calculations reported here the norm of the projected active orbitals is typically larger than 99.0% and never below 98.5% of the original crystal Wannier functions.

A. Correlation induced corrections to the diagonal matrix elements

Relaxation and polarization effects in the immediate neighborhood of an oxygen hole were computed by separate restricted open-shell HF (ROHF) calculations on [O₃₉Mg₃₀] clusters. The multiconfiguration MCSCF module of the MOLPRO package was employed for this purpose. The active region C_A of the $[O_{39}Mg_{30}]$ cluster contains a central $2p^5$ (or $2s^1$) O⁻ ion and four of the twelve nearest O²⁻ ligands. These four oxygen neighbors are all chosen to be in the same plane and we denote them as $O_{xy}^1,...O_{xy}^4$ (see Fig.4). In addition, we include in the cluster \mathcal{C} the NN cations and all the O's in the next coordination shell of each of the \mathcal{O}^i_{xy} sites, as sketched in Fig.4. These additional Mg and O neighbors represent the so-called buffer region \mathcal{C}_B and ensure an accurate description of the tails of the orbitals centered in the active region \mathcal{C}_A .

When performing the ROHF calculations, the oxygen hole orbital is kept frozen³⁹. We also freeze in our calculations the core-like 1s, 2s, and 2p shells of all Mg ions. Orbital relaxation effects are listed in Table I for both 2s

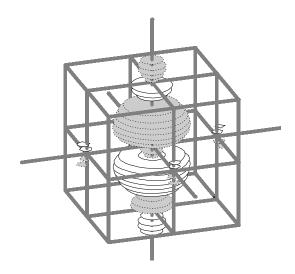


FIG. 3: Mg p-like conduction-band Wannier orbital after projection onto a $[{\rm Mg_{19}O_{14}}]$ cluster.

and 2p oxygen holes. The on-site relaxation effect is quite large, more than 2 eV. There is also a substantial relaxation/polarization effect associated with the first oxygen neighbors. For the four O_{xy}^i ligands included in the active region \mathcal{C}_A of the cluster, this effect amounts to about 0.41 eV in the presence of an O 2s hole and to 0.40 eVin the presence of a 2p hole. In Table I, we multiplied these numbers by three because there is a total of twelve oxygens in that coordination shell. We obtain thus a good estimate for the relaxation and polarization effects up to the nearest O neighbors. The fact that these relaxation/polarization effects are additive was checked by extra calculations with smaller, double-zeta basis sets⁵¹ on two different clusters: a [O₃₉Mg₃₀] cluster including only the four O_{xy}^i sites in the active region and a $[O_{55}Mg_{38}]$ cluster where all twelve first oxygen neighbors were allowed to polarize.

The four lowest-energy (N+1) conduction-band states imply Mg $3s^1$ and $3p^1$ electron configurations. ROHF calculations were performed for such configurations on a $[Mg_{19}O_{38}]$ cluster with a $[MgO_6]$ kernel as active region, \mathcal{C}_A . Beyond the $[MgO_6]$ kernel, this cluster incorporates again all Mg and O ions in the first two coordination shells of the active ligands. The on-site relaxation effects associated with the addition of an electron in a localized Mg 3s or 3p Wannier orbital are vanishingly small. The relaxation effects at the adjacent O sites induce energy shifts of 0.80–0.85 eV, see Table I. In these calculations the open-shell active orbitals $(Mg 3s^1 \text{ or } 3p^1)$ were again kept frozen³⁹. We note that the energetic effect is nearly the same for the $3s^1$ and $3p^1$ conduction-band states. At the scale of Fig.1 at least, it induces an uniform down-

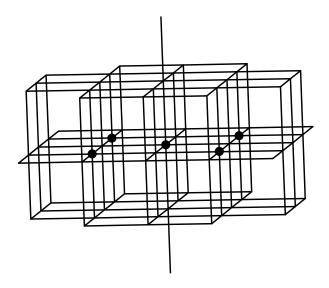


FIG. 4: Sketch of the $[O_{39}Mg_{30}]$ cluster employed for the calculation of short-range relaxation effects on the O valence-band states. The shortest line segments are Mg–O "bonds". The so-called active region \mathcal{C}_A includes a central O ion and four nearest oxygen neighbors O_{xy}^i in the "horizontal" plane, see text. These active ions are shown as small black spheres.

TABLE I: Correlation induced corrections to the diagonal Hamiltonian matrix elements for the valence-band O 2s, 2p and conduction-band Mg 3s, 3p states. All numbers are in eV. Negative corrections induce upwards shifts of the valence bands and shifts to lower energies for the conduction bands.

$\Delta H_{nn}(0)$	O~2s	O 2p	Mg 3s	Mg 3p
On-site orb. relaxation NN orb. relaxation Long-range polarization Total	-2.64 -1.23 -1.80 -5.67	-2.04 -1.20 -1.80 -5.04	-0.81 -2.25 -3.06	-0.84 -2.25 -3.09

wards shift of the center of gravity of the 3s-3p band complex.

The data listed in Table I indicate that the on-site orbital relaxation and relaxation and polarization effects at the nearest oxygen sites in the presence of an extra electron or extra electron hole results in a reduction of the HF band gap by about 4.05 eV, that is, more than 45% of the difference between the HF and experimental values. Large corrections are also expected to arise from long-range polarization effects. The long-range polarization energy of a dielectric due to the presence of an extra charge $\pm e$ can be expressed as $\Delta E(\infty) = \Delta E(R) - C/R^1$, where

$$C = \frac{\epsilon_0 - 1}{2\epsilon_0} e^2 \,, \tag{11}$$

 ϵ_0 is the static dielectric constant of the material, and R defines a sphere around the extra charge beyond which the dielectric response reaches its asymptotic value ϵ_0 . The energy increment $\Delta E(R)$ denotes the relaxation and polarization energy up to the radius R around the added particle. The constant C can be obtained by choosing two different radii R_1 and R_2 where the quantities $\Delta E(R_1)$ and $\Delta E(R_2)$ are computed, see for example Ref. 35. However, we adopt here a simpler approach. We calculate the corrections due to long-range polarization by using the experimental value for the static dielectric constant, $\epsilon_0 = 9.7$. Since relaxation and polarization effects related to the nearest oxygen neighbors were already accounted for, both, for the valence-band hole states and the conduction-band electrons (see Table I) and since the core-like electrons of the $\mathrm{Mg}^{2\dot{+}}$ ions can be ignored in these calculations, we set R as the average of the radii of the first and second oxygen coordination shells around a localized 2p hole or 3s(3p) electron: $R = (a\sqrt{2}/2 + a)/2$ for the O 2p valence-band states, where $a = 4.217 \,\text{Å}$ is the lattice constant⁵⁰, and $R = (a/2 + a\sqrt{3}/2)/2$ for the conduction-band states. The corrections to the diagonal matrix elements of the Hamiltonian are then

$$\Delta H_{nn}(\mathbf{0}) = -\frac{\epsilon_0 - 1}{2\epsilon_0} \frac{e^2}{R} , \qquad (12)$$

about -1.80 eV for the O 2s/2p bands and -2.25 eV for the Mg 3s/3p bands. These numbers are also included in Table I.

Before we discuss band narrowing (or broadening) due to correlation effects which also affect the band gap, we consider the loss of ground-state correlations. The latter leads again to a shift of the center of gravity of the bands. As pointed out in the previous section, some of the configurations that are present in the N-particle ground-state are blocked when an electron is added or removed. We investigated such correlations by CI calculations with single and double excitations (CISD) and discuss first differential effects for the N and (N-1) states. Since the oxygen valence-band Wannier orbitals are rather localized, we designed a cluster with a single O ion in the active region. Around this central O site we added one shell of Mg ions (6 Mg's) and two shells of anions (12+6 O's) to build the buffer region C_B . In the CISD calculations for the N and (N-1) configurations we correlate the 2s and 2p orbitals of the central O ion. Thereby the occupancy of the hole orbital is kept frozen in the calculations for the (N-1) states, which is referred to as the frozen hole approximation 38,39. Sets of separately optimized orbitals were used for the hole states of the (N-1)-particle system, as discussed above. We found that for a 2p hole the correction to the on-site matrix element of the Hamiltonian is $\Delta H_{nn}(\mathbf{0}) = 0.85 \text{ eV}$, i.e., the 2p valence bands are downshifted by 0.85 eV. For the O 2s hole states, this correction amounts to 0.99 eV. One would expect similarly an upwards shift of the conduction bands. However, the situation is somewhat different here. When an extra electron is attached to the Mg^{2+} ion, it polarizes the closed shells of the core. This is the dominant effect now because Mg²⁺ has no valence electrons. We may employ for our analysis the high-quality results obtained for a free Mg ion by Doll et al.⁵³. The correction to the ROHF Mg⁺ \rightarrow Mg²⁺ ionization potential was found to be 0.27 eV in Ref. 53. A similar differential correlation effect is occurring for the conduction-band states in bulk MgO. Therefore the conduction bands are shifted downwards instead of upwards and a partial cancellation between loss of ground-state correlation effects for the valence and conduction bands is taking place. The net result is a slight increase of the gap between the valence and conduction bands, in the range of 0.5 eV. Among the different contributions discussed here, this appears to produce the smallest corrections to the gap. More advanced calculations for studying such differential correlation effects are left for future work.

To summarize the results listed in Table I, relaxation and polarization effects in bulk MgO are responsible for a reduction of the Hartree-Fock gap by 8.1 eV, which represents about 95% of the difference between the HF and experimental values, 16.2 and 7.8, respectively. Improved agreement is expected between our results and the experimental data when applying higher-quality basis sets. We mention in this context that a reduction of 3.8 eV is obtained for the HF gap when going from valence double-zeta to triple-zeta basis sets. Nevertheless, this large energy difference is mainly related to the very poor representation of the conduction-band states in the

TABLE II: Nearest–neighbor (NN), $\mathbf{R}_{NN} = (1,1,0)a/2$, and next-nearest-neighbor (NNN), $\mathbf{R}_{NNN} = (1,0,0)a$, hopping matrix elements for the conduction-band Mg 3s and 3p orbitals, see text. Results of frozen-orbital CI (FO-CI) are listed in the second column; NOCI results in terms of separately optimized, relaxed orbitals (RO-NOCI) are given in the third column. All numbers are in eV.

$t_{nn'}(\mathbf{R})$	FO-CI	RO-NOCI	
t_{NN} :			
3s - 3s	0.41	0.42	
$3p_{x(y)} - 3p_{x(y)}$	0.66	0.69	
$3p_{x(y)} - 3p_{y(x)}$	0.72	0.77	
$3p_z - 3p_z$	0.13	0.13	
t_{NNN} :			
3s - 3s	0.36	0.37	
$3p_x - 3p_x$	0.77	0.74	
$3p_{y(z)} - 3p_{y(z)}$	0.13	0.12	

calculations with double-zeta basis functions and such effects will be less substantial by further extension of the basis sets. We also keep in mind that differential correlation effects due to the existence of a different number of electrons in the system's ground-state and in the $(N\pm 1)$ excited states determine a small correction in the opposite direction, i.e., a slight increase of the fundamental gap.

B. Off-diagonal matrix elements

We discuss next the effect of correlations on the widths of the different bands. For that purpose the off-diagonal matrix elements of the effective Hamiltonian (10) have to be determined, i.e., the so-called hopping terms. At the Hartree-Fock level, these matrix elements are obtained by solving 2×2 secular equations where both wavefunctions are expressed in terms of localized HF orbitals, see Eqs. (4), (5), and (8). Relaxation and polarization effects in the nearby surroundings of the added electron (or hole) are obtained by separate SCF optimizations for the $(N\pm 1)$ states. The separate optimization of the $(N\pm 1)$ wavefunctions leads to sets of non-orthogonal orbitals. There will be thus both Hamiltonian and overlap matrix elements between the extra electron (extra hole) wavefunctions $\Psi^{N\pm 1}_{0n\sigma}$ and $\Psi^{N\pm 1}_{\mathbf{R}n'\sigma}$. The calculation of such matrix elements has been recently implemented in MOLPRO⁵⁴. It is based on the transformation of the CI vectors to bi-orthogonal orbitals and follows an idea suggested by Malmqvist⁵⁵. A similar approach was developed by Broer et al. in Groningen⁵⁶. Instead of putting these matrix elements directly into the eigenvalue equations that determine the Bloch energies $\epsilon_n(\mathbf{k})$ we extract from these data effective hopping parameters associated with various pairs of orbitals. Comparison between such effective hopping terms and the HF off-diagonal Hamiltonian matrix elements offers an insightful picture of how correlation effects modify the inter-site interactions and

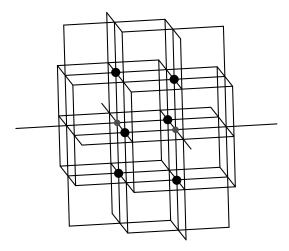


FIG. 5: Sketch of the $[Mg_{28}O_{36}]$ cluster employed for the calculation of the NN conduction-band hopppings. The active region C_A includes two NN Mg sites, small grey spheres, plus the bridging and apical ligands, large black spheres.

consequently the widths of the bands. For energetically degenerate states, the effective hopping is defined as

$$t_{nn'} = (H_{nn'} - S_{nn'}H_{nn})/(1 - S_{nn'}^2), \qquad (13)$$

where $H_{nn'}$ and $S_{nn'}$ are the Hamiltonian and overlap matrix elements between the $(N \pm 1)$ states n and n'. Since the separately optimized wavefunctions are expressed in terms of sets of non-orthogonal orbitals, this type of secular problem is usually referred to as non-orthogonal CI (NOCI). In the case of mutually orthogonal states $S_{nn'}$ is zero and $t_{nn'}(\mathbf{R}) = H_{nn'}(\mathbf{R})$.

Nearest-neighbor and next-nearest-neighbor (NNN) hopping matrix elements for the more diffuse conductionband Mg 3s and 3p orbitals are listed in Table II. In the rocksalt structure, there are two O ions bridging two NN cations, whereas two NNN Mg's share a single oxygen. We designed a [Mg₂₈O₃₆] cluster for calculating the NN hoppings and a [Mg₃₆O₄₇] cluster for the NNN matrix elements. As active regions, we employed [Mg₂O₆] and [Mg₂O₉] kernels, respectively, see Fig.5 and Fig.6. As in the calculations for the on-site matrix elements, for each of these clusters we included in the buffer region C_B all metal and O ions in the first two coordination shells of the active oxygens. Since the second cluster consists of 83 atoms, we reduced the computational effort by removing the polarization functions at the oxygen sites for this particular cluster. Two different values are given in Table II for each hopping matrix element. Results extracted from 2×2 CI calculations in terms of frozen HF Wannier orbitals are listed in the second column. In the third column, we allowed for full relaxation of the 2s and 2p orbitals of the O ions included in the active region \mathcal{C}_A , for each of the (N+1) configuration state functions entering the 2×2 CI. As illustrated in Figs. 5 and 6, those active anions are the ligand(s) bridging two Mg sites and the ligands which are nearest oxygen neighbors of one of the active Mg sites and also of the bridging ligand(s).

The results show that the separate optimization of the (N+1) wavefunctions induces only minor changes on the electron hoppings, in the range of few percent. An interesting feature is that variations occur in both directions, i.e., some of the effective hoppings are slightly enlarged by taking into account short-range relaxation and polarization effects and some are reduced. Since these changes are quite small, the width (and the overall structure) of the lower conduction bands will change very little. Regarding the longer-range polarization effects, we expect that their influence on the hopping terms is negligible.

In the case of non-equivalent orbitals, neighboring Mg 3s and Mg 3p, it is more difficult to define some NOCI effective hoppings and the corresponding data is not shown in Table II. Nevertheless, the effect of short-range relaxation and polarization is also small for these interactions, with reductions of the CI splittings of few meV.

The same type of analysis was applied to the (N-1) O $2s^{1}/2p^{5}$ hole states. The spatial extent of the oxygen orbitals and the inter-site matrix elements are significantly smaller. Since the largest relaxation effects concern orbitals at the same site, see Table I, we included in a first step only two ligands in the active regions of our clusters. Nevertheless, few other atomic shells were added around these active sites in the buffer region \mathcal{C}_B . We considered a $[O_{28}Mg_{10}]$ cluster for computing the relaxation effects on the NN hoppings and a $[O_{32}Mg_{11}]$ cluster for the NNN terms. The results are collected in Table III. The largest corrections arise for the NN matrix elements, with absolute values that are similar to the corrections obtained for the NN conduction-band hoppings. In relative numbers, these corrections are somewhat larger, by 10% to 20% for the NN hoppings. A very different situation occurs in strongly correlated oxides such as the layered cuprates, where the existence of an antiferromagnetic spin background determines a reduction of the effective quasiparticle hoppings by a factor of four⁵⁷.

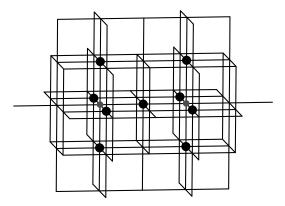


FIG. 6: Sketch of the $[Mg_{36}O_{47}]$ cluster employed for the calculation of the NNN conduction-band hopppings. The active region \mathcal{C}_A includes two Mg sites, small grey spheres, and nine O neighbors, large black spheres.

TABLE III: NN, $\mathbf{R}_{NN} = (1,1,0)a/2$, and NNN, $\mathbf{R}_{NNN} = (1,0,0)a$, hopping matrix elements for the valence-band states. Results of frozen-orbital CI (FO-CI) are listed in the second column; NOCI results in terms of separately optimized, relaxed orbitals (RO-NOCI) are given in the third column. All numbers are in eV. Data for the oxygen 2s orbitals are also included, although the O 2s band is much below the O 2p bands. The values in parentheses include orbital relaxation effects at four additional O sites, see text.

$t_{nn'}(\mathbf{R})$	FO-CI	RO-NOCI	
t_{NN} :			
2s-2s	0.10	0.12 (0.11)	
$2p_{x(y)} - 2p_{x(y)}$	0.32	0.37 (0.36)	
$2p_{x(y)} - 2p_{y(x)}$	0.42	0.49 (0.47)	
$2p_z - 2p_z$	0.12	0.14 (0.13)	
t_{NNN} :			
2s-2s	0.01	0.01	
$2p_x - 2p_x$	0.06	0.06	

The trends displayed in Table III are confirmed by multi-reference CISD [MRCI(SD)] calculations. We correlated in these calculations the eight 2s and 2p orbitals at the two NN oxygen sites and the reference active space included those two orbitals involved in the hopping process. The hopping matrix element is half of the energy separation between the lowest two eigenstates. The MRCI values for the $2p_x-2p_x$ and $2p_x-2p_y$ hoppings, for example, are 0.33 and 0.43 eV. These numbers are again larger than the HF results, although the magnitude of the effect is smaller as compared to the NOCI calculations. Having the data of the SCF calculations for the (N-1)oxygen hole states at hand, we performed an analysis of the changes produced in the composition of the (relaxed) orbitals in the immediate vicinity of an $O^ 2s^1$ or $2p^5$ anion. We found that an oxygen hole is causing polarization and "bending" of the 2p orbitals at the NN ligand sites. The "bending" of the NN 2p orbitals towards the O^- site takes place through both p-s and p_i-p_j mixing. These effects result in stronger inter-site orbital overlap and explain the fact that the effective NOCI hoppings are larger than the corresponding HF values. A similar analysis for the conduction-band (N+1) states is complicated by the presence of several sets of "active" orbitals, Mg 3s, 3p and bridging O 2p, and we could draw no clear conclusions in that case.

Extra calculations were performed for the NN hopping matrix elements with four additional oxygen sites included in the active region of the cluster, C_A . Those are the four ligands which are nearest neighbors to both O ions involved in the hopping process. They are situated in the median plane of the segment $\mathbf{R}_{NN} = (1,1,0)a/2$. A cluster composed of 62 sites, $[O_{40}\mathrm{Mg}_{22}]$, was employed for these calculations. The results are given in Table III in parentheses. The corrections due to relaxation and polarization at the four nearest O neighbors are very small, 0.01 to 0.02 eV. The effect of these corrections is to reduce somewhat the absolute values of the NN hoppings.

Experimental studies for characterizing the valence electronic structure of MgO have been carried out using x-ray photoelectron spectroscopy (XPS)⁵⁸ and angle-resolved ultraviolet photoelectron spectroscopy (ARUPS)⁵⁹. The measured width of the O 2p bands is about $6.5~{\rm eV}^{58,59}$. The HF valence-band width is $5.50~{\rm eV}$ and inclusion of local correlations leads to a slight broadening of the O 2p bands, which brings the ab initio result in good agreement with the experiment. For comparison, density-functional calculations within the local density approximation and using the same Gaussian basis sets as in the HF calculations predict a width of $4.68~{\rm eV}$ for the O 2p bands.

The correlation induced corrections to the widths of the bands also influence the band gap. In the fcc lattice the dispersion of p bands at the Γ point depends on two of the nearest-neighbor hoppings, $\epsilon_x(\Gamma) = const. + 8t_{x,x}^{(110)} - 4t_{x,x}^{(011)} + \dots^{60,61}$. With the notations from Table III, $t_{x,x}^{(011)} = t_{z,z}^{(110)}$. Corrections of 0.04 eV for $t_{x,x}^{(110)}$ and 0.01 eV for $t_{x,x}^{(011)}$ (or $t_{z,z}^{(110)}$), see Table III, imply an upwards shift of the O 2p bands at the Γ point and a narrowing of the fundamental gap by about 0.30 eV. For the Mg 3s-3p conduction-band complex such changes at the Γ point are smaller because the correlation induced corrections to the 3s-3s and 3s-3p inter-site matrix elements are lower.

IV. SUMMARY AND CONCLUSIONS

We have analyzed the different correlation contributions to the energy gap of MgO and to the widths of the conduction and valence bands. This was done within the quasiparticle description. As regards correlation effects we have distinguished between relaxation and polarization around an electron or hole added to the (neutral) ground-state. The net result is a reduction of the Hartree-Fock gap from 16.2 eV to a value of 8.1 eV. This has to be compared with a measured energy gap of 7.8 eV. Within the local density approximation (LDA) to density functional theory a gap of 5.0 eV is found. This is not surprising since LDA is known to produce too small gaps for insulators.

The calculations were performed with triple-zeta basis sets. Since on the Hartree-Fock level the calculated gap differs for double- and triple-zeta basis sets by 3.8 eV, one may consider the good agreement with the experimental gap as somewhat fortuitous. Triple-zeta basis sets are known, however, to produce reliable results in quantum chemistry and therefore a further extension of the basis set should keep the corrections small. Presently we are not able to work with larger basis sets.

It was shown that a large contribution to the correlation induced corrections to the fundamental gap comes from on-site and nearest-neighbor relaxation, i.e., from the immediate neighborhood of the added particle. But also the long-range part of the polarization generated by the extra particle contributes substantially to the reduction of the HF gap. This long-range part can be treated in a continuum approximation thereby using the known dielectric constant of MgO. The so-called loss of ground-state correlations makes a small contribution in MgO. The reason is that the conduction-band Wannier orbitals have predominant Mg 3s or 3p character. The added electron will essentially go thus to a Mg site where it polarizes the closed $1s^2$, $2s^2$, and $2p^6$ shells. This effect reduces the gap and counteracts the loss of ground-state correlations which occurs when an electron is removed (hole state) and therefore can no longer contribute to the correlations of the remaining ones. Finally, also changes in the widths of the bands influence the energy gap.

One surprising effect which we found is an enhancement of the width of the valence bands when local corre-

lations are taken into account. This is the opposite one expects since correlations lead usually to band narrowing instead of broadening. It results from a slight bending of nearest-neighbor 2p orbitals towards a ${\rm O}^-$ site when we account for correlations. The bending of the nearest-neighbor ligand orbitals increases the wavefunction overlap and hence facilitates the hopping.

The present results prove the usefulness of band calculations based on quantum chemical techniques. They allow for well controlled approximations and a transparent interpretation of the different microscopic processes which determine the size of the energy gap and the widths of the bands.

We thank Dr. E. Pahl and Prof. J. Fink for fruitful discussions.

- ¹ P. Fulde, Electron Correlations in Molecules and Solids (Springer-Verlag, Berlin, 1995).
- N. H. March, Electron Correlation in Molecules and Condensed Phases (Plenum Press, New York, 1996).
- ³ P. Fulde, Adv. Phys. **51**, 909 (2002).
- ⁴ R. McWeeny, Methods of Molecular Quantum Mechanics (Academic Press, London, 1996).
- ⁵ T. Helgaker, P. Jørgensen, and J. Olsen, Molecular Electronic-Structure Theory (Wiley, Chichester, 2000).
- ⁶ S. Suhai, Phys. Rev. B **27**, 3506 (1983).
- ⁷ C.-M. Liegener, J. Chem. Phys. **88**, 6999 (1988).
- ⁸ J. J. Ladik, Phys. Rep. **313**, 171 (1999).
- ⁹ J.-Q. Sun and R. J. Bartlett, J. Chem. Phys. **104**, 8553 (1996).
- ¹⁰ S. Hirata and S. Iwata, J. Chem. Phys. **109**, 4147 (1998).
- ¹¹ P. Reinhardt and J.-P. Malrieu, J. Chem. Phys. **109**, 7632 (1998).
- ¹² P. Y. Ayala, K. N. Kudin, and G. E. Scuseria, J. Chem. Phys. **115**, 9698 (2001).
- ¹³ A. Abdurahman, A. Shukla, and M. Dolg, Phys. Rev. B 65, 115106 (2002).
- ¹⁴ C. Pisani, M. Busso, G. Capecchi, S. Casassa, R. Dovesi, L. Maschio, C. Zicovich-Wilson, and M. Schütz, J. Chem. Phys. **122**, 094113 (2005).
- ¹⁵ S. Hirata and R. J. Bartlett, J. Chem. Phys. **112**, 7339 (2000).
- ¹⁶ R. Pino and G. E. Scuseria, J. Chem. Phys. **121**, 2553 (2004).
- ¹⁷ M. Albrecht and J. Igarashi, J. Phys. Soc. Jpn. **70**, 1035 (2001).
- ¹⁸ M. Albrecht and P. Fulde, Phys. Stat. Sol. B **234**, 313 (2002).
- ¹⁹ C. Buth, U. Birkenheuer, M. Albrecht, and P. Fulde, Phys. Rev. B **72**, 195107 (2005).
- ²⁰ W. Förner, Int. J. Quantum Chem. **43**, 221 (1992).
- ²¹ W. Förner, R. Knab, J. Čížek, and J. Ladik, J. Chem. Phys. **106**, 10248 (1997).
- ²² K. Fink and V. Staemmler, J. Chem. Phys. **103**, 2603 (1995).
- ²³ S. Hirata, R. Podeszwa, M. Tobita, and R. J. Bartlett, J. Chem. Phys. **120**, 2581 (2004).
- ²⁴ C. M. Zicovich-Wilson, R. Dovesi, and V. R. Saunders, J.

- Chem. Phys. **115**, 9708 (2001).
- ²⁵ U. Birkenheuer and D. Izotov, Phys. Rev. B **71**, 125116 (2005).
- ²⁶ A. Shukla, M. Dolg, P. Fulde, and H. Stoll, Phys. Rev. B 57, 1471 (1998).
- ²⁷ A. Shukla, M. Dolg, P. Fulde, and H. Stoll, Phys. Rev. B 60, 5211 (1999).
- ²⁸ A. Abdurahman, A. Shukla, and M. Dolg, J. Chem. Phys. 112, 4801 (2000).
- ²⁹ P. Sony and A. Shukla, Phys. Rev. B **73**, 165106 (2006).
- ³⁰ J. Rubio, A. Povill, J. P. Malrieu, and P. Reinhardt, J. Chem. Phys. **107**, 10044 (1997).
- ³¹ H. Stoll, Phys. Rev. B **46**, 6700 (1992).
- ³² For a review, see B. Paulus, Phys. Rep. **428**, 1 (2006).
- ³³ S. Horsch, P. Horsch, and P. Fulde, Phys. Rev. B **29**, 1870 (1984).
- ³⁴ J. Gräfenstein, H. Stoll, and P. Fulde, Chem. Phys. Lett. 215, 611 (1993).
- ³⁵ J. Gräfenstein, H. Stoll, and P. Fulde, Phys. Rev. B 55, 13588 (1997).
- ³⁶ M. Albrecht, P. Fulde, and H. Stoll, Chem. Phys. Lett. 319, 355 (2000).
- ³⁷ V. Bezugly and U. Birkenheuer, Chem. Phys. Lett. **399**, 57 (2004).
- ³⁸ U. Birkenheuer, P. Fulde, and H. Stoll, Theor. Chem. Acc. 116, 398 (2006).
- ³⁹ E. Pahl and U. Birkenheuer, J. Chem. Phys. **124**, 214101 (2006).
- ⁴⁰ E. Pahl, U. Birkenheuer, H. Stoll, and P. Fulde, manuscript in preparation.
- ⁴¹ V. R. Saunders, R. Dovesi, C. Roetti, M. Causà, N. M. Harrison et al., CRYSTAL 2000, University of Torino, Italy.
- ⁴² U. Birkenheuer, C. Willnauer, M. von Arnim, W. Alsheimer, and D. Izotov, Scientific Report, Max-Planck-Institut für Physik komplexer Systeme, Dresden, 2002.
- ⁴³ U. Birkenheuer, manuscript in preparation.
- ⁴⁴ C. Roetti, R. Dovesi, M. von Arnim, W. Alsheimer, and U. Birkenheuer, the CRYSTAL-MOLPRO interface, Max-Planck-Institut für Physik komplexer Systeme, Dresden, Germany.
- ⁴⁵ P. Pulay, Chem. Phys. Lett. **100**, 151 (1983); S. Saebø and P. Pulay, Chem. Phys. Lett. **113**, 13 (1985).
- ⁴⁶ H.-J. Werner, P. J. Knowles, R. Lindh, F. R. Manby, M.

- Schütz $et\ al.,\ {\tt MOLPRO}\ 2006,\ {\tt Cardiff}\ {\tt University},\ {\tt United}\ {\tt Kingdom}.$
- ⁴⁷ S. S. P. Parkin, C. Kaiser, A. Panchula, P. M. Rice, B. Hughes, M. Samant, and S.-H. Yang, Nature Materials 3, 862 (2004); S. Yuasa, T. Nagahama, A. Fukushima, Y. Suzuki, and K. Ando, Nature Materials 3, 868 (2004).
- ⁴⁸ S. Choopun, R. D. Vispute, W. Yang, R. P. Sharma, T. Venkatesan, and H. Shen, Appl. Phys. Lett. **80**, 1529 (2002).
- ⁴⁹ R. C. Whited, C. J. Flaten, and W. C. Walker, Solid State Commun. **13**, 1903 (1973).
- ⁵⁰ S. Sasaki, K. Fujino, and Y. Takeuchi, Proc. Japan. Acad. B **55**, 43 (1979).
- M. I. McCarthy and N. M. Harrison, Phys. Rev. B 49, 8574 (1994); R. Dovesi, C. Roetti, C. Freyria-Fava, E. Aprá, V. R. Saunders, and N. M. Harrison, Philos. Trans. R. Soc. London Ser. A 341, 203 (1992).
- ⁵² J. Pipek and P. G. Mezey, J. Chem. Phys. **90**, 4916 (1989).

- ⁵³ K. Doll, M. Dolg, P. Fulde, and H. Stoll, Phys. Rev. B **52**, 4842 (1995).
- ⁵⁴ A. O. Mitrushchenkov and H.-J. Werner, Mol. Phys., in press (2007).
- ⁵⁵ P.-Å. Malmqvist, Int. J. Quantum Chem. **30**, 479 (1986).
- ⁵⁶ R. Broer and W. C. Nieuwpoort, Chem. Phys. **54**, 291 (1981); Theor. Chim. Acta **73**, 405 (1988).
- ⁵⁷ L. Hozoi, S. Nishimoto, and C. de Graaf, Phys. Rev. B, in press (2007) [also available on cond-mat/0611720].
- ⁵⁸ S. P. Kowalczyk, F. R. McFeely, L. Ley, V. T. Gritsyna, and D. A. Shirley, Solid State Commun. 23, 161 (1977).
- ⁵⁹ L. H. Tjeng, A. R. Vos, and G. A. Sawatzky, Surf. Sci. 235, 269 (1990).
- ⁶⁰ J. C. Slater and G. F. Koster, Phys. Rev. **94**, 1498 (1954).
- ⁶¹ N. W. Ashcroft and N. D. Mermin, Solid State Physics (Thomson Brooks/Cole, 1976).